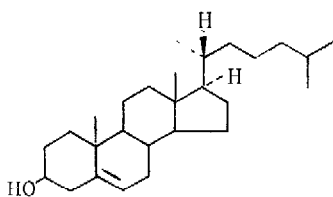


AN APPROACH TO THE STEREOCONTROLLED CREATION OF AN  
ACYCLIC SIDE CHAIN OF SOME NATURAL PRODUCTS

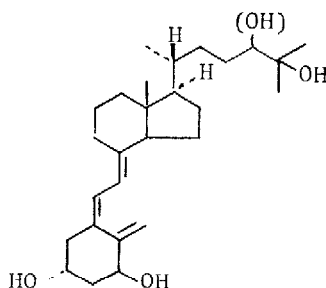
Barry M. Trost, Douglas F. Taber, and Joseph B. Alper  
Department of Chemistry  
University of Wisconsin  
Madison, Wisconsin 53706

(Received in USA 1 June 1976; received in UK for publication 7 September 1976)

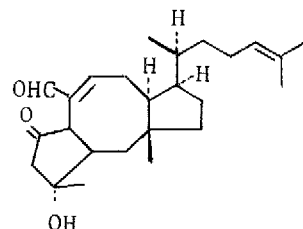
The acyclic C<sub>7</sub> side chain of steroids 1, Vitamin D metabolites 2, ophiobolin 3, etc. require the development of methods for the selective control of the stereochemistry of the acyclic carbon bearing methyl relative to ring



1



2



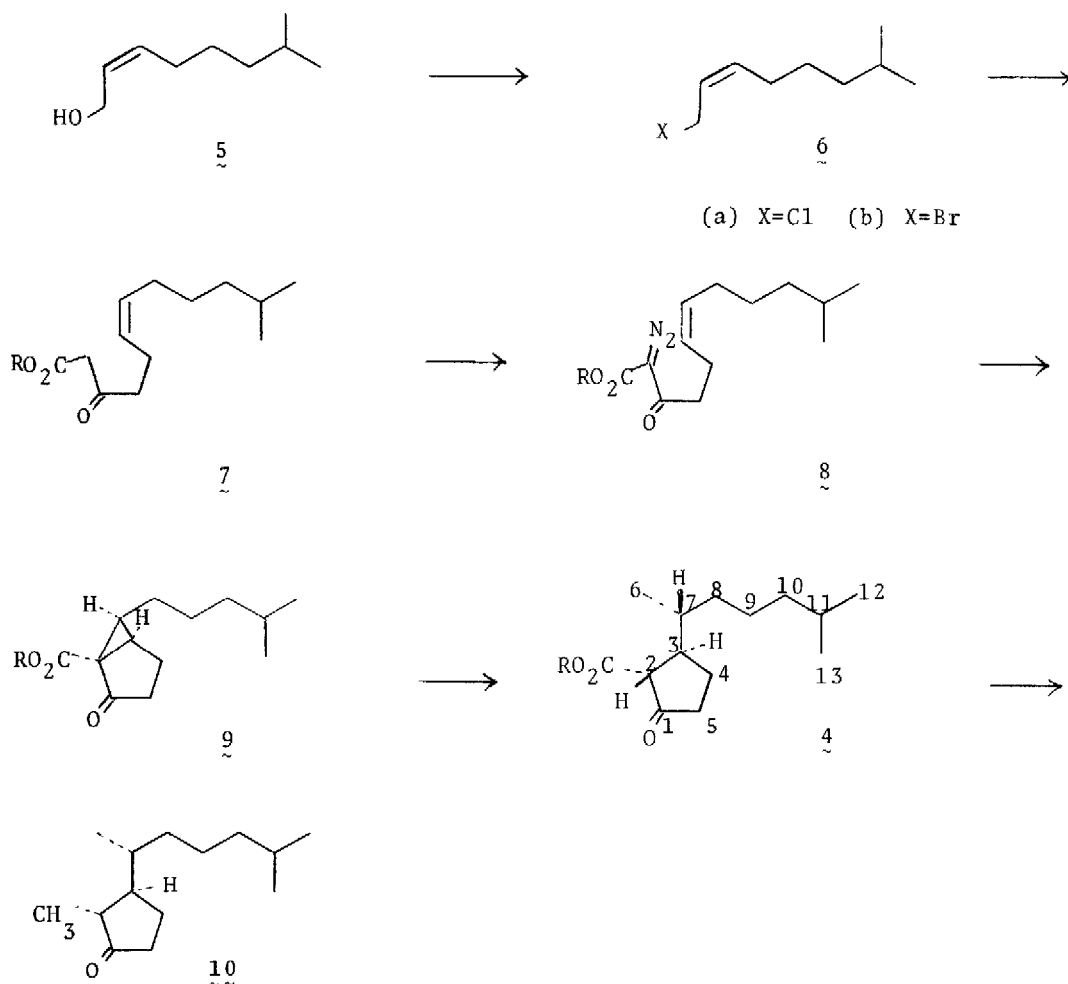
3

geometry.<sup>1</sup> We wish to report an approach to this problem based upon the opening of a cyclopropane ring with an organocopper reagent.<sup>2,3</sup> The target compound 4 can be envisioned as an intermediate in the total synthesis of the above types of compound.

The scheme outlines the approach. 7-Methyloct-2-en-1-ol (5), available from 2-2-buten-1,4-diol as previously described,<sup>4</sup> was converted to the chloride 6a<sup>5a</sup> [NCS,<sup>6</sup> (CH<sub>3</sub>)<sub>2</sub>S, CH<sub>2</sub>Cl<sub>2</sub>, 0°, bp 90° @ 11 mm, 76%] or preferably the bromide 6b<sup>5b</sup> [PBr<sub>3</sub>, CaH<sub>2</sub>,<sup>7</sup> ether, 0°, bp 54-7° @ 1.4 mm, 80%]. Alkylation of the dianion of methyl acetoacetate generated in the usual way<sup>8</sup> [alkylation, 25°, bp 109° @ 0.1 mm, 71%] gave the unsaturated ketoester 7<sup>5a</sup> [IR 1752, 1726, 1659, 1640 cm<sup>-1</sup>; NMR δ 5.16-6.48, m, 2H; 1.0-1.7, m, 5H; 0.90, d, J=7, 6H]. The corresponding t-butyl ester was also easily prepared in an identical fashion

utilizing *t*-butyl acetoacetate. Diazo transfer<sup>9</sup> [ $\text{TsN}_3$ ,  $\text{CH}_3\text{CN}$ ,  $(\text{C}_2\text{H}_5)_3\text{N}$ ,  $25^\circ$ ] was monitored by the appearance of strong IR bands at 2140, 1720, and  $1657\text{ cm}^{-1}$

Scheme. Synthetic Approach to 4.



and the crude product **8** subjected directly to intramolecular cyclization<sup>10</sup> [Copper powder,  $\text{PhCH}_3$ , reflux, 73-80%] to give the desired cyclopropane **9**<sup>5a</sup> [IR  $1740\text{ cm}^{-1}$ , NMR  $\delta$  3.68, s, 3H; 1.75-2.6, m, 6H; 1.1-1.65, m, 7H; 0.88, d,  $J=7\text{ Hz}$ , 6H]. It is interesting to note that **9** behaves as if it is exceptionally polar on tlc.

The key reaction employs the ring cleavage of the cyclopropane with cuprates.<sup>2,3</sup> Indeed, treatment of **9** with lithium dimethylcuprate [ $0^\circ$ , ether, 86%]

gave the desired 4<sup>5a</sup> [IR 1764, 1736, 1660, 1612  $\text{cm}^{-1}$ ; NMR  $\delta$  3.64, s, 3H; 2.76 d,  $J = 11$  Hz, 1H; 1.0-2.6, m, 4H; 1.0-1.8, m, 9H; 0.88, d,  $J = 7$  Hz, 9H] which was homogeneous by spectral and chromatographic criteria. The E stereochemistry of the ring is suggested by the 11 Hz coupling constant between the H on C(2) and C(3) and by thermodynamics. The stereochemistry of C(7) relative to the ring junction follows from the cis stereochemistry of the olefin and the known inversion of the configuration in the cuprate ring opening.<sup>2c,3</sup> The compound was further characterized by methylation, hydrolysis, and decarboxylation of 4 ( $R = \text{CO}_2\text{C}_4\text{H}_9\text{t}$ ) to give 10.<sup>5a</sup> This approach should be applicable to either stereochemistry at C(7) since the E olefin should give the opposite configuration.

Acknowledgement: We wish to thank the National Institute of Health and the National Science Foundation for their generous support of our work.

#### References

1. For a quite different approach to this type of problem see B. M. Trost and T. R. Verhoeven, J. Amer. Chem. Soc., 98, 630 (1976); J. Ficini, J. d'Angelo, and J. Noire, ibid., 96, 1213 (1974).
2. For references to opening of cyclopropyl rings with organocuprates see a) H. O. House and P. D. Weeks, J. Amer. Chem. Soc., 97, 2778 (1975); b) P. A. Grieco and R. Finkelhor, J. Org. Chem., 38, 2100 (1973); c) E. J. Corey and P. L. Fuchs, J. Amer. Chem. Soc., 94, 4014 (1972); d) W. G. Dauben and W. M. Welch, Tetrahedron Letters, 4531 (1971); e) G. Daviaud and Ph. Miginiac, ibid., 997 (1972); J. A. Marshall and R. A. Ruden, J. Org. Chem., 37, 659 (1972).
3. A similar ring opening to that reported here was published while this work was in progress. R. D. Clark and C. H. Heathcock, Tetrahedron Letters, 529 (1975).
4. J. Cologne and G. Poilane, Bull. Soc. Chim. France, 1953 (1955).
5. a) This compound was fully characterized by spectral means and elemental composition. b) This compound was fully characterized by spectral means.
6. E. J. Corey, C. U. Kim, and M. Takeda, Tetrahedron Letters, 4339 (1972).

7. E. J. Corey, D. E. Cane, and L. Libit, J. Amer. Chem. Soc., 93, 7016 (1971).
8. S. N. Huckin and L. Weiler, J. Amer. Chem. Soc., 96, 1082 (1974).
9. M. Regitz, J. Hocker, and A. Liedhegener, Org. Syn. Coll. Vol. V., 179 (1973).
10. C. G. Stork and J. Ficini, J. Amer. Chem. Soc., 83, 4678 (1961).